



Açaí (*Euterpe oleraceae*) 'BRS Pará': A tropical fruit source of antioxidant dietary fiber and high antioxidant capacity oil

Maria do Socorro M. Rufino^{a,b,1}, Jara Pérez-Jiménez^{b,2}, Sara Arranz^b, Ricardo Elesbão Alves^{c,*}, Edy S. de Brito^c, Maria S.P. Oliveira^d, Fulgencio Saura-Calixto^b

^a Federal Rural University of the Semi-Arid, BR 110, Km 47, Presidente Costa e Silva, 59625-900, Mossoró, RN, Brazil

^b Department of Metabolism and Nutrition, Institute for Food Science and Technology and Nutrition (ICTAN-CSIC), Calle José Antonio Novais, 10, 28040 Madrid, Spain

^c Embrapa Tropical Agroindustry, R. Dra. Sara Mesquita, 2270, Pici, 60511-110, Fortaleza, CE, Brazil

^d Embrapa Western Amazonia, Trav. Dr. Enéas Pinheiro s/n°, 66095-100, Belém, PA, Brazil

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ABSTRACT

This article reports a study of the concentrations of dietary fiber (DF) and antioxidant capacity in fruits (pulp and oil) of a new açaí (*Euterpe oleraceae*) cultivar—'BRS-Pará', with a view to determine the possibility of using it as a source of antioxidants in functional foods or dietary supplements. Results show that 'BRS-Pará' açaí fruits has a high content of DF (71% dry matter) and oil (20.82%) as well as a high antioxidant capacity in both defatted matter and oil. 'BRS-Pará' Açaí fruits can be considered as an excellent source of antioxidant dietary fiber. Antioxidant capacity of açaí 'BRS-Pará' oil by DPPH assay was higher ($EC_{50} = 646.3$ g/g DPPH) than extra virgin olive oil ($EC_{50} = 2057.27$ g/g DPPH). These features provide açaí 'BRS-Pará' fruits with considerable potential for nutritional and health applications.

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1. Introduction

Açaí (*Euterpe oleraceae*), also known as cabbage palm, is a tropical species which bears a dark purple, berry-like fruit, clustered into bunches. Its exportation to other non-tropical countries, to be used mainly in fruit juices, has increased during several years.

Recently, much attention has been paid to its antioxidant capacity and its possible role as a functional food or food ingredient (Coisson, Travaglia, Piana, Capasso, & Arlorio, 2005; Jensen et al., 2008; Mertens-Talcott et al., 2008; Pozo-Insfran, Percival, & Talcott, 2006; Ribeiro et al., 2010; Schreckinger, Lotton, Lila, & Mejia, 2010). The phytochemical and

Abbreviations: ABTS, 2,2'-Azino-bis(3-ethylbenzo-thiazoline-6-sulfonic acid); ADF, antioxidant dietary fiber; AE, antiradical efficiency ($AE = 1/(EC_{50} \cdot t_{EC50})$); DF, dietary fiber; DPPH[•], 2,2-Diphenyl-1-picrylhydrazyl; EC_{50} , concentration of antioxidant needed to reduce the original amount of radical by 50%; EPP, extractable polyphenols; FRAP, Ferric Reducing/Antioxidant Power; GAE, gallic acid equivalents; IDF, insoluble dietary fiber; NSP, nonstarch polysaccharides; ORAC, Oxygen Radical Absorbance Capacity; SDF, soluble dietary fiber; t_{EC50} , time needed to reach the steady state to EC_{50} concentration; TPITZ, 2,4,6-Tris(2-pyridyl)-s-triazine; Trolox, 6-Hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid.

* Corresponding author. Tel.: +55 85 3391 7202; fax: +55 85 3391 7222.

E-mail address: elesbao@cnpat.embrapa.br (R.E. Alves).

¹ Current address: Food Technology Department, Federal University of Ceara, Av. Mister Hall, 2977, Bloco 858, Pici, 60356-000, Fortaleza, CE, Brazil.

² Current address: Institute of Advanced Chemistry of Catalonia-CSIC, c/ Jordi Girona, 18-26, 08034 Barcelona, Spain.

nutrient composition of açaí have been investigated (Rufino, Alves, Brito, Silveira, & Moura, 2009; Rufino et al., 2010). Anthocyanins, proanthocyanidins, and other flavonoids were found to be the major phytochemicals in freeze-dried açaí (Schauss et al., 2006) and some works have dealt with the antioxidant capacity of açaí pulp (Souza et al., 2009; Rufino, Alves, Brito, Perez-Jimezes, & Saura-Calixto, 2009; Rufino, Fernandes, Alves, & Brito, 2009; Rufino et al., 2010;). However, since açaí is an oil-rich fruit, it may be necessary, when determining its antioxidant capacity, to study separately its oil and its defatted fraction to avoid interferences, as it has been suggested for similar fruits (Arranz, Pérez-Jiménez, & Saura-Calixto, 2008). This has not been carried out up to the moment.

Cereals are usually studied as main sources of dietary fiber (DF). Nevertheless, it is well-known that the DF from some fruits, that contains a higher proportion of soluble dietary fiber (SDF) and associated bioactive compounds than cereals, has properties related to gastrointestinal health and prevention of chronic diseases (Spiller, 2001). Antioxidant DF (ADF) is defined as a natural product that combines the beneficial effects of DF and natural antioxidants, such as polyphenol compounds (Saura-Calixto, 1998). On the one hand, ADF can be used as a dietary supplement to improve gastrointestinal health and to prevent cardiovascular diseases (Pérez-Jiménez, Serrano, et al., 2008), and on the other as an ingredient in seafood and meat products to prevent lipid oxidation (Sánchez-Alonso, Jiménez-Escrig, Saura-Calixto, & Borderías, 2006).

Traditionally açaí grows on flooded areas. However, through a genetic breeding program based on phenotypic selection from its germplasm

bank, Embrapa Western Amazonia (Belém, PA, Brazil) developed a cultivar—the ‘BRS-Pará’—suitable for growing on stable land, as a result the production system of this plant has now been modified making it easier and more productive than the traditional system.

The aim of this work was to study the concentrations of DF and antioxidant capacity in açai, ‘BRS-Pará’ with a view to determining the possibility of using it as a source of antioxidant in functional foods or dietary supplements. Due to its high oil content, antioxidant capacity was studied separately in the defatted pulp and in the oil, to avoid interferences. Finally, since the association of antioxidants with DF may produce specific physiological effects, the polyphenols and the antioxidant capacity associated to DF were also determined.

2. Materials and methods

2.1. Chemicals

Pepsin (2000 FIP-U/g), glucose, inositol and *N*-methylimidazole were obtained from Merck (Darmstadt, Germany). Amyloglucosidase (14 IU/mg) was from, Roche, Mannheim, Germany. Pancreatin, α -amylase (17.5 IU/mg), 2,2'-Azino-bis(3-ethylbenzo-thiazoline-6-sulfonic acid) (ABTS), 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox), catechin, gallic acid, galacturonic acid, galactose and mannose were obtained from Sigma-Aldrich Química, S.A. (Madrid, Spain). 2,4,6-Tris(2-pyridyl)-*s*-triazine (TPTZ) was from Fluka Chemicals (Madrid, Spain). Dinitrosalicylic acid, 3,6'-dihydroxy-spiro-[isobenzofuran-1-[3H],9[9H]-xanthen]-3-one (fluorescein) and iron III-chlorure-6-hydrate were from Panreac, Castellar del Vallés (Barcelona, Spain). All reagents used were of analytical grade.

2.2. Samples

Fruits of açai ‘BRS-Pará’ were harvested at Embrapa Western Amazonia at Belém-PA, Brazil. After harvesting, the fruits were transported to the Postharvest Physiology and Technology Laboratory, at Embrapa Tropical Agroindustry, Fortaleza-CE, Brazil. Two kilograms of fruit were harvested in the second semester of 2007 at the commercial maturity stage (completely ripe). Samples were taken

from 10 different trees and from different regions of them, to achieve a homogeneous sample. They were processed in a domestic blender (Walita, Brazil) to obtain a pulp and the seeds were discarded. Pulp was later divided in three sub-samples to carry out the different analysis and each one of them was freeze-dried (LH 4500, Terroni Fauvel, Brazil) and milled to a particle size of less than 0.5 mm in a centrifugal mill. The fact that these fruits were from the same genotype (clone) guaranteed a similarity in the chemical composition of them.

2.3. Methods

Previously the samples were defatted with petroleum ether at 60 °C on a Soxhlet apparatus, using two extraction cycles of 30 min. Figs. 1 and 2 show a scheme of the treatments performed to the samples to determine DF and antioxidant capacity.

Determinations were performed in triplicate and reported on a dry matter. Results are expressed as mean values \pm standard deviation.

2.3.1. Dietary fiber determination

The DF was measured based on the procedure described by Saura-Calixto, Garcia-Alonso, Goñi and Bravo (2000). This method combines enzymatic treatments and separation of digestible compounds by dialysis using physiological conditions (temperature and pH), obtaining the fraction of food that is not digested (Fig. 1). Total DF was calculated as the sum of insoluble dietary fiber or IDF (constituted by nonstarch polysaccharides or NSP, Klason lignin, resistant protein, ash, extractable polyphenols, proanthocyanidins, and hydrolyzable tannins) and soluble dietary fiber or SDF (constituted by soluble nonstarch polysaccharides or NSP and extractable polyphenols).

Samples (300 mg) were incubated with pepsin (0.2 mL of a 300 mg/mL solution in 0.08 M HCl-KCl buffer, pH 1.5, 40 °C, 1 h), pancreatin (1 mL of a 5 mg/mL solution in 0.1 M phosphate buffer, pH 7.5, 37 °C, 6 h) and α -amylase (1 mL of a 120 mg/mL solution in 0.1 M Tris-maleate buffer, pH 6.9, 37 °C, 16 h). Samples were centrifuged (15 min, 3000 \times g) and supernatants removed. Residues were washed twice with 5 mL of distilled water, and all supernatants were combined. Each supernatant was incubated with 100 μ L of amyloglucosidase for 45 min at 60 °C before being transferred to dialysis tubes

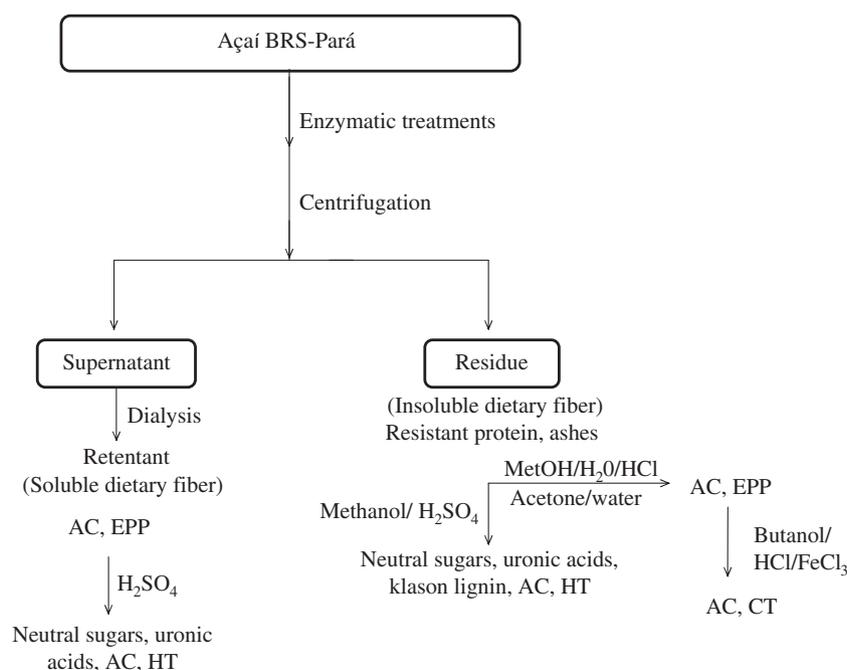


Fig. 1. Flow chart showing determination of dietary fiber and associated antioxidants in açai ‘BRS-Pará’. AC: antioxidant capacity; EPP: extractable polyphenols; HT: hydrolysable tannins; CT: condensed tannins.

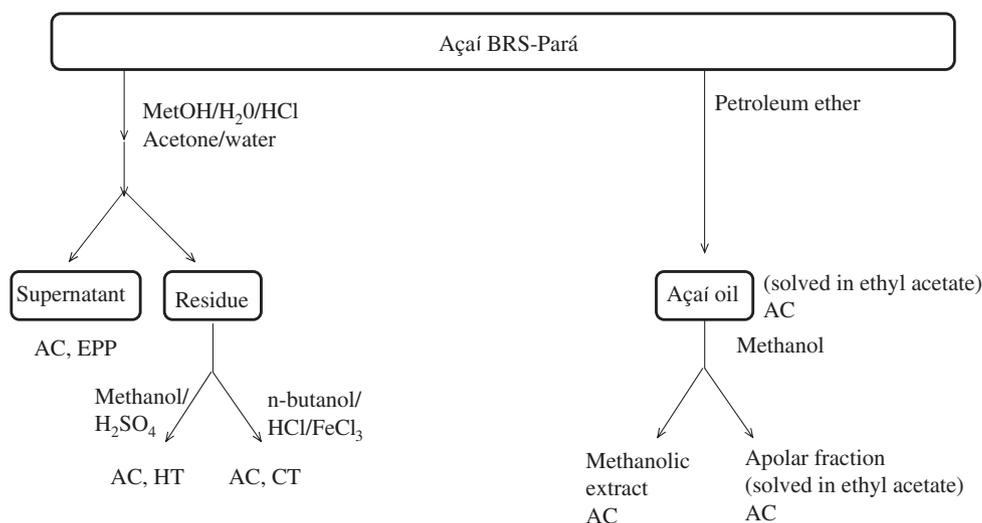


Fig. 2. Flow chart showing determination of antioxidant capacity of the defatted sample and oil of açai 'BRS-Pará'. AC: antioxidant capacity; EPP: extractable polyphenols; HT: hydrolysable tannins; CT: condensed tannins.

(12,000–14,000 molecular weight cutoff, Visking dialysis tubing; Medicell International Ltd., London, U.K.) and dialyzed against water for 48 h at 25 °C to eliminate digestible compounds. The products of all these treatments were therefore a residue after enzymatic treatments, corresponding to IDF, and a supernatant of enzymatic treatments later subjected to dialysis, corresponding to SDF.

In SDF, soluble NSP were hydrolyzed with 1 M sulfuric acid at 100 °C for 90 min and measured as the sum of neutral sugars, determined by GC, and uronic acids, determined spectrophotometrically (Scott, 1979) using galacturonic acid as standard. Regarding GC, neutral sugars were derivatized to alditol acetates (Englyst, Wiggins & Cummings, 1984) by a first treatment with NH_4OH , octan-2-ol and NaBH_4 during 30 min at 40 °C, followed by a second treatment with methylimidazole and acetic anhydride during 15 min and a final addition of KOH. A Shimadzu GC-14A chromatograph (Shimadzu Co., Kyoto, Japan) fitted with a flame ionization detector and a SP-2330 capillary column (30 m \times 0.32 mm i.d., catalog no. 2-4073, Supelco, Bellefonte, PA) were used. Analytical conditions were as follows: column temperature, 240 °C (isothermal); injector temperature, 270 °C; detector temperature, 270 °C; carrier gas, nitrogen. Inositol was used as internal standard. Finally, extractable polyphenols were also determined in SDF as described in the later part—see Section 2.3.2.3.

Regarding IDF, the residue was weighed to determine gravimetrically IDF content in the sample, and it was divided in three fractions to analyze its different constituents. A first fraction was used to determine insoluble NSP and Klason lignin (Southgate, 1969): after treatment with sulphuric acid (12 M, 20 °C for 3 h; dilution to 1 M and incubation for 2 h, 100 °C), insoluble NSP were determined as the sum of neutral sugars and uronic acids as described above for soluble NSP, and Klason lignin was determined gravimetrically. In the supernatant of this treatment hydrolyzable tannins content was also determined—as discussed after Section 2.3.2.3. A second fraction of the residue was used to determine in it resistant protein and ash—see 2.3.3. The third fraction was subjected to a first treatment to determine extractable polyphenols—see Section 2.3.2.1, followed by a second treatment to determine condensed tannins content—see Section 2.3.2.3.

2.3.2. Antioxidant capacity and phenolic compounds determination

2.3.2.1. Extraction of antioxidants. 0.5 g of either IDF of açai (Fig. 1) or defatted açai pulp (Fig. 2) was placed in a capped centrifuge tube;

20 mL of acidic methanol/water (50:50, v/v; pH 2) was added and the tube was thoroughly shaken at room temperature for 1 h. The tube was centrifuged at $2500 \times g$ for 10 min and the supernatant recovered. Twenty milliliters of acetone/water (70:30, v/v) were added to the residue, and shaking and centrifugation repeated. Methanolic and acetic extracts were combined and used to determine the antioxidant capacity associated with extractable antioxidants (Figs. 1 and 2). The residues of these extractions were subjected either to hydrolysis with H_2SO_4 to release hydrolyzable tannins (Figs. 1 and 2) or to treatment with n-butanol/HCl/ FeCl_3 to release anthocyanins from proanthocyanidins or condensed tannins (Figs. 1 and 2)—see conditions in Section 2.3.2.3. Antioxidant capacity was determined in both hydrolyzable tannins and condensed tannins.

Total antioxidant capacity was determined directly in vegetable oils, after diluting aliquots in ethyl acetate. To determine separately antioxidant capacity associated to polar and apolar compounds, 5 mL of oil were mixed with 5 mL of methanol. The mixture was vigorously stirred for 20 min and centrifuged at $2500 \times g$ for 10 min and the supernatant was recovered. Another 5 mL were added and the same process was repeated. Antioxidant capacity was measured directly in the methanolic extract (that extracts polar compounds) and in the remaining oil (apolar fraction), after dilution with ethyl acetate (Espín, Soler-Rivas, & Wichers, 2000).

2.3.2.2. Antioxidant capacity methods

2.3.2.2.1. DPPH[•] (free-radical scavenging) assay. It was used the method described by Brand-Williams, Cuvelier, and Berset (1995), later modified by Sánchez-Moreno, Larrauri, and Saura-Calixto (1998), in order to determine kinetic parameters. After adjusting the blank with methanol, 0.1 mL of the sample was mixed with 3.9 mL of a DPPH[•] methanolic solution (60 μM). The absorbance at 515 nm was measured until the reaction reached the plateau. A calibration curve at that wavelength was made to calculate the remaining DPPH[•]. The parameter EC_{50} , which reflects 50% depletion of DPPH[•], was expressed in terms of grams of açai equivalent per gram of DPPH[•] in the reaction medium. The time taken to reach the steady state at EC_{50} ($t_{\text{EC}_{50}}$) and the antiradical efficiency ($\text{AE} = 1/\text{EC}_{50} \cdot t_{\text{EC}_{50}}$) were also determined.

2.3.2.2.2. ABTS assay at a fixed end-point. ABTS radical cation ($\text{ABTS}^{•+}$) was produced by reacting 7 mM ABTS stock solution with 2.45 mM potassium persulfate and allowing the mixture to stand in the dark at room temperature for 12–16 h before use. The $\text{ABTS}^{•+}$ solution was

diluted with methanol to an absorbance of 0.70 ± 0.02 at 658 nm. After the addition of 100 μ L of sample or Trolox standard to 3.9 mL of diluted ABTS⁺ solution, absorbance readings were taken every 20 s, using a Beckman DU-640 (Beckman Instruments Inc. Fullerton, CA, USA) spectrophotometer. The reaction was monitored during 6 min. The percentage inhibition of absorbance versus time was plotted, and the area below the curve (0–6 min) was calculated (Re et al., 1999). Methanolic solutions of known Trolox concentrations were used for calibration.

2.3.2.2.3. ABTS assay expressed kinetically. The ABTS radical cation is generated as described for the ABTS assay at a fixed end-point. A recent procedure described modified the original method so as to determine kinetic parameters. An aliquot of the sample extract (0.1 mL) is added to 3.9 mL of ABTS⁺ (0.044 g/L) in methanol which was prepared daily. Absorbances at 658 nm are measured at different time intervals on a spectrophotometer until the reaction reaches a plateau. The ABTS⁺ concentration in the reaction medium is calculated by plotting concentration vs. absorbance. EC₅₀, t_{EC50} and AE are calculated as in the DPPH assay (Pérez-Jiménez & Saura-Calixto, 2008).

2.3.2.2.4. FRAP (ferric reducing antioxidant power) assay. 900 μ L of FRAP reagent (80% acetate buffer, 10% TPTZ 10 mM and 10% iron III-chlorure-6-hydrate 20 mM), freshly prepared and warmed at 37 °C, was mixed with 90 μ L of distilled water and either 30 μ L of test sample or standard or appropriate reagent blank. Reading at the absorption maximum (595 nm) was taken every 15 s, using a spectrophotometer. The readings at 30 min were selected for calculation of FRAP values (Benzie & Strain, 1996; Pulido, Bravo, & Saura-Calixto, 2000). Solutions of known Trolox concentrations were used for calibration.

2.3.2.2.5. ORAC (oxygen radical absorbance capacity) assay. Sample/blank is mixed with PBS buffer, AAPH and fluorescein. Fluorescence was recorded until it reached zero (excitation wavelength 493 nm, emission wavelength 515 nm) in a fluorescence spectrophotometer Perkin-Elmer LS 55 at 37 °C (Ou, Hampsch-Woodill, & Prior, 2001). Results were calculated using the differences of areas under the fluorescein decay curve between the blank and the sample and were expressed as Trolox equivalents.

2.3.2.3. Antioxidant compounds content. The content of extractable polyphenols (EPP) was determined in SDF, extracts of IDF (Fig. 1) and in extracts of the defatted pulp (Fig. 2) according to the Folin-Ciocalteu method (Singleton, Orthofer, & Lamuela-Raventós, 1999). Test sample (0.5 mL) was mixed with 1 mL of Folin-Ciocalteu reagent and swirled. After 3 min, 10 mL of sodium carbonate solution (75 g/L) was added and mixed. Additional distilled water was mixed thoroughly by inverting the tubes several times. After 1 h, the absorbance at 750 nm was recorded. The results were expressed as g of gallic acid equivalents (GAE)/100 g.

Proanthocyanidins (condensed tannins) not extracted by the previous aqueous-organic procedure were measured at 555 nm after hydrolysis with n-butanol/HCl/FeCl₃ (3 h, 100 °C) (Reed, McDowell, Van Soest, & Horvath, 1982) either in the residue of the extraction of IDF (Fig. 1) or in the residue of the extraction of açai defatted pulp (Fig. 2). Results were compared with carob pod (*Ceratonia siliqua* L.) proanthocyanidin standard (Nestlé, Ltd., Vers-Chez-les Blancs, Switzerland).

Hydrolyzable tannins were measured according to a method previously described (Hartzfeld, Forkner, Hunter, & Hagerman, 2002) by hydrolysis with methanol and sulfuric acid for 20 h at 85 °C. They were determined in the hydrolysates of IDF (Fig. 1) and in the residues of the extraction of açai defatted pulp (Fig. 2). Concentration was estimated by the Folin-Ciocalteu method (Singleton et al., 1999) and expressed as g GAE/100 g.

2.3.3. Proximate composition determination

Protein was determined using an automated nitrogen analyser FP-2000®; Dumas Leco Corp. Lipids were determined using a Soxhlet System HT extractor with petroleum ether, and fatty acid composition by GC, after derivatization to methyl sters (Gómez-Cortés et al., 2008).

Ash content was gravimetrically determined by incinerating samples in an electric muffle furnace at 550 °C for 16 h.

3. Results and discussion

3.1. Proximate composition

Proximate composition values are presented in Table 1. The most significant aspect of açai 'BRS-Para' is its high DF content (70% dry matter), the bulk of it being insoluble DF. Açai DF content, as in other tropical fruits, is much higher than in common fruits like apples, oranges or bananas in which it ranges from 17 to 36% (Saura-Calixto et al., 2000). Moreover, açai contains comparable levels of DF to other products described as rich in DF, such as pineapple shell or grape pomace (Larrauri, Rupérez, & Saura-Calixto, 1997; Pérez-Jiménez, Arranz, et al., 2008), and higher than other tropical fruits like guava or papaya (Jiménez-Escrig, Rincón, Pulido, & Saura-Calixto, 2001; Mahattanatawee et al., 2006).

Table 2 shows the composition of açai 'BRS-Pará' DF, including neutral sugars (determined individually), uronic acids, Klason lignin, resistant protein, and ash. Glucose and galactose were the main neutral sugars in SDF, while arabinose and xylose were the major monosaccharides in IDF indicating the presence of arabinoxylans.

These portions of non digestible carbohydrates reach the colon where they are potentially fermentable by colonic microbiota. Some of the metabolites generated during colonic fermentation of carbohydrates, as short chain fatty acids (especially butyrate), were described as beneficial for intestinal health (Wong, Souza, Kendall, Emam, & Jenkins, 2006).

Açai also has high oil content (Table 1). Analysis of fatty acids (Table 3) showed that this fruit is a source of fatty acids of potential nutritional interest; containing almost as much oleic acid as olive oil, which contains 70% oleic acid on average, and more than other oil rich sources such as soy, corn and sunflower. This is in line with results reported by other authors on açai from the Amazon estuary (Schauss et al., 2006).

Açai also possesses a high phenolic content that presumably contributes to its antioxidant capacity, as it is discussed in the later part. Finally, the protein, soluble sugars and mineral content agreed with other reported data (Menezes, Torres, & Srur, 2008).

3.2. Polyphenols and antioxidant capacity

The antioxidant capacity of any food sample comes from the combined synergic action of a mixture of compounds, including phenolics, carotenoids, vitamins C and E, etc. Except for certain fruits such as acerola (Alves, Chitarra, & Chitarra, 1995), in which vitamin C is one of the main component, in fruits like açai which contain relatively little of this vitamin, polyphenols are the main contributors to antioxidant capacity.

Table 1

Proximate composition of Açai 'BRS-Pará' fruit pulp.

Component	g/100 g dm ^a
Protein	6.27 ± 0.31
Ashes	1.99 ± 0.17
Soluble sugars	7.93 ± 2.11
Total lipids	20.82 ± 1.60
Soluble dietary fiber	2.75 ± 0.16 ^b
Insoluble dietary fiber	68.49 ± 1.21 ^c
Total dietary fiber	71.22 ± 1.22
Polyphenols ^d	
Extractable polyphenols	1.50 ± 0.05
Hydrolyzable tannins	1.59 ± 0.18
Condensed tannins	1.24 ± 0.14

dm: dry matter. Moisture: 85.7%.

^a Mean value ± standard deviation, n = 3.

^b Determined as non-starchy polysaccharides + associated polyphenols.

^c Determined gravimetrically.

^d A fraction of them is included in dietary fiber.

Table 2
Composition of dietary fiber of Açai 'BRS-Pará' pulp (% dm^a).

Component	Soluble dietary fiber	Insoluble dietary fiber	Total dietary fiber
Total neutral sugars	0.65 ± 0.04	11.90 ± 0.07	12.55 ± 0.10
Arabinose	0.10 ± 0.04	0.65 ± 0.02	0.75 ± 0.04
Fucose	n.d.	0.09 ± 0.01	0.09 ± 0.01
Galactose	0.16 ± 0.04	0.29 ± 0.01	0.45 ± 0.04
Glucose	0.39 ± 0.03	0.25 ± 0.01	0.64 ± 0.03
Mannose	0.003 ± 0.0001	0.03 ± 0.004	0.03 ± 0.005
Rhamnose	n.d.	n.d.	n.d.
Xylose	n.d.	10.59 ± 0.23	10.59 ± 0.23
Uronic acids	0.98 ± 0.08	15.96 ± 0.48	16.92 ± 0.70
Klason lignin	–	39.27 ± 0.02	39.27 ± 0.02
Resistant protein	–	5.60 ± 0.20	5.60 ± 0.20
Ash	–	1.02 ± 0.30	1.02 ± 0.30
Polyphenols	1.12 ± 0.13	2.93 ± 0.11	4.05 ± 0.17
Dietary fiber	2.75 ± 0.16	76.68 ± 0.80	79.43 ± 0.81

dm: dry matter.

^a Mean value ± standard deviation, n = 3. n.d. non detected.

Polyphenol content determined in the pulp aqueous–organic extracts can be seen in Table 4. The values are similar to those reported by other authors (Schauss et al., 2006). The main phenolic compounds reported in açai pulp have been cyanidin 3-O-glucoside, a cyanidin 3-O-rutinoside, homoorientin, orientin, and isovitexin (Gallori, Bilia, Bergonzi, Barbosa, & Vincieri, 2004), and in the related fruit jussara (*Euterpe edulis*) cyanidin 3-O-glucoside and cyanidin 3-O-rutinoside (Brito et al., 2007).

Total polyphenols present in the residue of these extractions that is, hydrolyzable and condensed tannins were determined, giving values of 1.59 and 1.24 g/100 g dry weight, respectively, being therefore as abundant as extractable polyphenols. Although non-bioavailable in the small intestine, these non extractable polyphenols reach the colon intact and there becomes fermentable substrates for colonic bacteria. The fermentation of these compounds release antioxidant metabolites that may improve the colonic status and yield some absorbable metabolites (Cerdá, Periago, Espín, & Tomás-Barberán, 2005; Gonthier et al., 2003).

The antioxidant capacity associated with these non-extractable phenolic compounds in açai had not been previously determined. Regarding the oil, it was found only one reference in which the antioxidant capacity of açai oil was measured by ORAC assay (Pacheco-Palencia, Mertens-Talcott, & Talcott, 2008). Therefore, another aim of this work was to determine the total antioxidant capacity of this fruit. In order to perform these determinations, it was necessary to defat the açai since its high oil content could interfere in the measurement of this parameter (Arranz et al., 2008).

3.2.1. Antioxidant capacity of defatted pulp

Antioxidant capacity associated with phenolic compounds was determined by FRAP, ABTS, DPPH and ORAC (Table 4)—the different pros and cons linked to each one of the available antioxidant capacity

Table 3
Fatty acid composition and oil of Açai 'BRS-Pará' fruit pulp.

Component	Fatty acid (g/100 g dm)	Oil (%)
Saturated	6.9	26.7
C16:0	5.3	25.3
C18:0	1.6	1.4
Monounsaturated	13.0	62.3
Cis-9 C16:1	1.1	5.4
Cis-9 C18:1	10.9	52.1
Cis-11 C18:1	1.0	4.8
Polyunsaturated	2.3	11.1
Cis-9, cis-12 C18:2	2.2	10.6
Cis-9, cis-12, cis-15 C18:3	0.1	0.5

dm: dry matter.

Table 4
Polyphenols and antioxidant capacity in aqueous–organic extracts and its residues of defatted Açai 'BRS-Pará' fruit pulp.^a

Analysis	Extractable polyphenols	Hydrolyzable tannins	Condensed tannins
Polyphenols (mg/g dm)	15.0 ± 0.5	15.9 ± 1.8	12.4 ± 1.4
FRAP (μmol Trolox/g dm)	128.44 ± 8.51	109.87 ± 5.19	n.d.
ORAC (μmol Trolox/g dm)	379.97 ± 11.96	1514.46 ± 20.20	n.d.
ABTS at a fixed end-point (μmol Trolox/g dm)	55.79 ± 1.12	20.73 ± 0.04	24.62 ± 4.52
ABTS expressed kinetically			
EC ₅₀ (g dm/g ABTS)	1.20 ± 0.04	13.61 ± 0.15	14.36 ± 0.59
t _{EC50} (min)	24.23 ± 1.58	25.20 ± 1.16	51.92 ± 7.03
AE ^b	0.034	0.003	0.001
DPPH			
EC ₅₀ (g dm/g DPPH)	10.20 ± 0.10	4.92 ± 0.10	n.d.
t _{EC50} (min)	41.43 ± 0.90	11.00 ± 0.69	n.d.
AE ^b	0.002	0.018	n.d.

dm: dry matter. n.d. non determined. All values are expressed per gram of dry whole açai.

^a Mean value ± standard deviation, n = 3.^b Antiradical efficiency, AE = 1/(EC₅₀ t_{EC50}).

assays make necessary the use of at least two of these techniques (Pérez-Jiménez, Arranz, et al., 2008). The aqueous–organic extracts obtained from defatted açai, showed high antioxidant capacity, mainly due to extractable polyphenols, as well as the residues of açai defatted pulp, where hydrolyzable and condensed tannins are the main contributors to antioxidant capacity. In the case of condensed tannins, only ABTS was a suitable procedure to determine their antioxidant capacity, since the other methods are not suitable to be performed in the butanol media needed the release of anthocyanins from proanthocyanidins.

Regarding kinetic measurements, extractable polyphenols were better radical scavengers than hydrolyzable tannins when ABTS assay was used (lower EC₅₀), but worse radical scavengers towards DPPH assay (EC₅₀ of 4.92 g/g for hydrolyzable tannins vs 10.20 for extractable polyphenols), indicating different mechanisms of reactions in these two assays (Huang, Ou, & Prior, 2005). Fig. 3 shows the kinetic behaviour of açai extracts vs DPPH[•].

3.2.2. Antioxidant capacity of the oil

Açai 'BRS-Pará' oil antioxidant capacity results are shown in Table 5. Only total oil and apolar fraction values can be directly compared, since both were measured using ethyl acetate as solvent, while the polar fraction measurements were performed in methanol. The EC₅₀ of the apolar fraction (536.5 g/g) was higher than that of total oil (646.3) indicating a higher antioxidant capacity; therefore, it seems that most antioxidants in açai oil are of an apolar nature, and the extraction of this fraction works as a mechanism to concentrate antioxidants.

We compared these values with extra virgin olive oil as a model of a lipophilic antioxidant-rich sample (Arranz et al., 2008), since oil extraction and antioxidant capacity measurements were equivalent in both studies. The EC₅₀ of total oil was lower in the açai oil than in the extra virgin olive oil (646.30 g/g DPPH vs. 2057.27 g/g DPPH) indicating higher antioxidant capacity in the former. In contrast, the antiradical efficiency (AE) of extra virgin olive oil was higher, because the time required for açai oil to achieve the EC₅₀ was longer than that required for the extra virgin olive oil. For the apolar fraction, again açai oil had a lower EC₅₀ than olive oil (536.53 g oil/g DPPH vs. 1210.96 g oil/g DPPH) and consequently higher antioxidant capacity to capture DPPH radical. Açai and olive oil apolar fractions showed similar results for t_{EC50}. However, regarding polar fraction, olive oil exhibited a much higher antioxidant capacity (EC₅₀ of 10.2 g/g) than açai oil (EC₅₀ of 1249.9 g/g).

Therefore, açai oil appears as a new source of lipophilic antioxidants, with an antioxidant activity similar to olive oil, known by its high antioxidant compounds content.

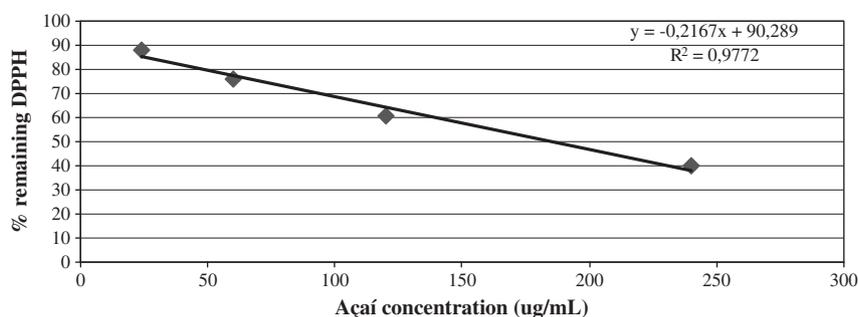


Fig. 3. Percentage of remaining DPPH• vs concentration of açai in the DPPH assay.

3.2.3. Polyphenols and antioxidant capacity associated with dietary fiber

DF, measured as indigestible fraction (Saura-Calixto et al., 2000), is composed of two fractions: a soluble fraction (supernatant from enzymatic digestion) and an insoluble fraction (residue from enzymatic digestion). Antioxidant capacity and phenolic compounds associated with DF were determined in both fractions (Table 6).

The SDF contained associated (in an aqueous buffer) extractable polyphenols, while the IDF contained associated (in an aqueous organic solvent) extractable polyphenols, hydrolyzable tannins and condensed tannins. These constitute an important fraction of the polyphenols present in açai pulp (Table 1), that is, most of them are associated with DF.

Although the values in Table 6 cannot be directly compared with those in Table 4, since the extraction methods were different (aqueous–organic vs enzymatic extraction), it can be seen that açai DF exhibits considerable associated antioxidant capacity. For example, the ABTS value of condensed tannins in the IDF was 20.51 ± 0.13 μmol Trolox/g dm, while hydrolyzable tannins in IDF gave an ORAC value of 155.61 ± 8.40 μmol Trolox/g dm. For comparative purposes, the antioxidant capacity associated with DF in a mixture of the fruits consumed in the Spanish diet has been calculated to be 2 μmol Trolox/g by ABTS assay (Serrano, Goñi, & Saura-Calixto, 2007). Moreover, the antioxidant capacity of açai DF is of nutritional significance, since these antioxidant compounds are thought to reach the colon intact in which they can produce a variety of beneficial effects.

In summary, antioxidant capacity and dietary fiber were determined in the fruit pulp of açai 'BRS-Pará'. Pulp contains polyphenols with high antioxidant capacity, most of them being associated with DF. Its oil has higher antioxidant capacity than olive oil and a comparable fatty acid profile. The high DF content of açai pulp and its associated polyphenols make this fruit a suitable source of antioxidant DF which may be used as a food ingredient to prevent lipid oxidation

Table 5

Antioxidant capacity of açai 'BRS-Pará' fruit pulp oil and extra virgin olive oil (DPPH method)^a.

	Total antioxidant capacity ^b	Polar fraction ^c	Apolar fraction ^d
Açai BRS Pará oil			
EC ₅₀ (g oil/g DPPH)	646.30 ± 38.40	1249.97 ± 32.39	536.53 ± 11.78
t _{EC50} (min)	35.69 ± 1.80	17.87 ± 0.33	37.25 ± 3.75
AE ^e	0.4×10^{-4}	0.4×10^{-4}	1.0×10^{-4}
Extra virgin olive oil^f			
EC ₅₀ (g oil/g DPPH)	2057.27 ± 75.56	10.25 ± 0.21	1210.96 ± 44.09
t _{EC50} (min)	4.86 ± 0.05	43.92 ± 0.86	35.31 ± 1.65
AE ^e	1×10^{-4}	2.2×10^{-4}	0

^a Mean value ± standard deviation, n = 3.

^b Determined in oil solved in ethyl acetate.

^c Antioxidant capacity determined in methanolic extract.

^d Antioxidant capacity determined in the remaining oil after methanolic extraction.

^e Antiradical Efficiency, AE = $1/(EC_{50} \cdot t_{EC50})$.

^f Arranz et al., 2008.

in seafood and meat products, as well as in dietary supplements. These data suggest that açai could have considerable potential for nutritional and health purposes.

Authors' responsibilities

The authors' responsibilities were as follows—F.S.C. and R.E.A., planned and monitored the work; M.S.M.R., J.P.-J. and S.A., conducted the experimental work; M.S.P.O., provided the samples to carry out the work and contributed to their preparation; M.S.M.R., J.P.-J. and R.E. A., wrote the first draft of the manuscript; and all authors contributed to writing the manuscript and approved the final version. None of the authors had a personal or financial conflict of interest.

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Table 6

Polyphenols and antioxidant capacity associated to dietary fiber of Açai 'BRS-Pará' fruit pulp.^a

Analysis	Soluble dietary fiber		Insoluble dietary fiber	
	Extractable polyphenols	Extractable polyphenols	Hydrolyzable tannins	Condensed tannins
Polyphenols (mg/g dm)	11.2 ± 1.3	4.2 ± 0.2	5.7 ± 0.5	19.4 ± 1.0
FRAP (μmol Trolox/g dm)	46.82 ± 4.65	11.91 ± 0.92	11.64 ± 0.84	n.d.
ORAC (μmol Trolox/g dm)	603.03 ± 9.18	53.72 ± 6.65	155.61 ± 8.40	n.d.
ABTS at a fixed end-point (μmol Trolox/g dm)	78.16 ± 0.49	3.96 ± 0.22	n.d.	20.51 ± 0.13
ABTS expressed kinetically				
EC ₅₀ (g dm/g ABTS)	8.95 ± 0.04	67.58 ± 2.11	n.d.	14.41 ± 0.06
t _{EC50} (min)	10.00 ± 0.80	20.47 ± 2.04	n.d.	26.87 ± 0.49
AE ^b	0.011	0.001	n.d.	0.002
DPPH				
EC ₅₀ (g dm/g DPPH)	39.43 ± 0.10	119.20 ± 5.42	47.31 ± 1.16	n.d.
t _{EC50} (min)	23.11 ± 0.06	31.22 ± 0.58	7.99 ± 0.92	n.d.
AE ^b	0.001	0.0003	0.003	n.d.

^a Mean value ± standard deviation, n = 3. d. m. dry matter. n.d. non determined.

^b Antiradical Efficiency, AE = $1/(EC_{50} \cdot t_{EC50})$.

References

- Alves, R. E., Chitarra, A. B., & Chitarra, M. I. F. (1995). Postharvest physiology of Acerola (*Malpighia emarginata* D.C.) fruits: Maturation changes, respiratory activity and refrigerated storage at ambient and modified atmospheres. *Acta Horticulturae*, 370, 223–230.
- Arranz, S., Pérez-Jiménez, J., & Saura-Calixto, F. (2008). Antioxidant capacity of walnut (*Juglans regia* L.): Contribution of oil and defatted matter. *European Food Research and Technology*, 227, 425–431.
- Benzie, I. F. F., & Strain, J. J. (1996). The ferric reducing ability of plasma (FRAP) as a measure of antioxidant power: The FRAP assay. *Analytical Biochemistry*, 239, 70–76.
- Brand-Williams, W., Cuvelier, M. E., & Berset, C. (1995). Use of a free radical method to evaluate antioxidant activity. *Lebensmittel Wissenschaft und Technologie*, 28, 25–30.
- Brito, E. S., Araújo, M. C. P., Alves, R. E., Carkeet, C., Clevidence, B. A., & Novotny, J. A. (2007). Anthocyanins present in selected tropical fruits: Acerola, jambolão, jussara, and guajiru. *Journal of Agricultural and Food Chemistry*, 55, 9389–9394.
- Cerdá, B., Periago, P., Espín, J. C., & Tomás-Barberán, F. A. (2005). Metabolism of antioxidant and chemopreventive ellagitannins from strawberries, raspberries, walnuts, and oak-aged wine in humans: Identification of biomarkers and individual variability. *Journal of Agricultural and Food Chemistry*, 53, 227–235.
- Coisson, J. D., Travaglia, F., Piana, G., Capasso, M., & Arlorio, M. (2005). *Euterpe oleracea* juice as a functional pigment for yogurt. *Food Research International*, 38, 893–897.
- Englyst, H. H., Wiggins, H. S., & Cummings, J. H. (1984). Determination of the non-starch polysaccharides in plant foods by gas liquid chromatography of constituents sugars alditol acetates. *The Analyst*, 109, 937–942.
- Espín, J. C., Soler-Rivas, C., & Wichers, H. J. (2000). Characterization of the total free radical scavenger capacity of vegetable oils and oil fractions using 2,2-diphenyl-1-picrylhydrazyl radical. *Journal of Agricultural and Food Chemistry*, 48, 648–656.
- Gallori, S., Bilia, A. R., Bergonzi, M. C., Barbosa, W. L. R., & Vincieri, F. F. (2004). Polyphenolic constituents of fruit pulp of *Euterpe oleracea* Mart. (açai palm). *Chromatographia*, 59, 739–743.
- Gómez-Cortés, P., Frutos, P., Mantecón, A. R., Juárez, M., Fuente, M. A., & Hervás, G. (2008). Addition of olive oil to dairy ewe diets: Effect on milk fatty acid profile and animal performance. *Journal of Dairy Science*, 91, 3119–3127.
- Gonthier, M. P., Donovan, J. L., Texier, O., Felgines, C., Remesy, C., & Scalbert, A. (2003). Metabolism of dietary procyanidins in rats. *Free Radicals Biology & Medicine*, 35, 837–844.
- Hartzfeld, P. W., Forkner, R., Hunter, M. D., & Hagerman, A. E. (2002). Determination of hydrolysable tannins (gallotannins and ellagitannins) after reaction with potassium iodate. *Journal of Agricultural and Food Chemistry*, 50, 1785–1790.
- Huang, D., Ou, B., & Prior, R. L. (2005). The chemistry behind antioxidant capacity assays. *Journal of Agricultural and Food Chemistry*, 53, 1841–1856.
- Jensen, G. S., Wu, X., Patterson, K. M., Barnes, J., Carter, S. G., Scherwitz, L., et al. (2008). *In Vitro* and *In Vivo* antioxidant and anti-inflammatory capacities of an antioxidant-rich fruit and berry juice blend. Results of a pilot and randomized, double-blinded, placebo-controlled, crossover study. *Journal of Agricultural and Food Chemistry*, 56, 8326–8333.
- Jiménez-Escrig, A., Rincón, M., Pulido, R., & Saura-Calixto, F. (2001). Guava fruit (*Psidium guajava* L.) as a new source of antioxidant dietary fiber. *Journal of Agricultural and Food Chemistry*, 49, 5489–5493.
- Larrauri, J. A., Rupérez, P., & Saura-Calixto, F. (1997). Effect of drying temperature on the stability of polyphenols and antioxidant activity of red grape pomace peels. *Journal of Agricultural and Food Chemistry*, 45, 1390–1393.
- Mahattanatawee, K., Manthey, J. A., Luzio, G., Talcott, S. T., Goodner, K., & Baldwin, E. A. (2006). Total antioxidant activity and fiber content of select Florida-grown tropical fruits. *Journal of Agricultural and Food Chemistry*, 54, 7355–7363.
- Menezes, E. M. S., Torres, A. T., & Srur, A. U. S. (2008). Valor nutricional da polpa de açai (*Euterpe oleracea* Mart) liofilizada. *Acta Amazônica*, 38, 311–316.
- Mertens-Talcott, S. U., Rios, J., Jilma-Stohlawetz, P., Pacheco-Palencia, L. A., Meibohm, B., Talcott, S. T., et al. (2008). Pharmacokinetics of anthocyanins and antioxidant effects after the consumption of anthocyanin-rich açai juice and pulp (*Euterpe oleracea* Mart.) in human healthy volunteers. *Journal of Agricultural and Food Chemistry*, 56, 7796–7802.
- Ou, B., Hampsch-Woodill, M., & Prior, R. L. (2001). Development and validation of an improved oxygen radical absorbance capacity assay using fluorescein as the fluorescent probe. *Journal of Agricultural and Food Chemistry*, 49, 4619–4626.
- Pacheco-Palencia, L. A., Mertens-Talcott, S., & Talcott, S. T. (2008). Chemical composition, antioxidant properties, and thermal stability of a phytochemical enriched oil from açai (*Euterpe oleracea* Mart.). *Journal of Agricultural and Food Chemistry*, 56, 4631–4636.
- Pérez-Jiménez, J., Arranz, S., Taberner, M., Díaz-Rubio, M. E., Serrano, J., Goñi, I., et al. (2008). Updated methodology to determine antioxidant capacity in plant foods, oils and beverages: Extraction, measurement and expression of results. *Food Research International*, 41, 274–285.
- Pérez-Jiménez, J., & Saura-Calixto, F. (2008). Anti-oxidant capacity of dietary polyphenols determined by ABTS assay: A kinetic expression of the results. *International Journal of Food Science & Technology*, 43, 185–191.
- Pérez-Jiménez, J., Serrano, J., Taberner, M., Arranz, S., Díaz-Rubio, M. E., García-Diz, L., et al. (2008). Effects of grape antioxidant dietary fiber on cardiovascular disease risk factors. *Nutrition*, 24, 646–653.
- Pozo-Insfran, D. D., Percival, S. S., & Talcott, S. T. (2006). Açai (*Euterpe oleracea* Mart.) polyphenolics in their glycoside and aglycone forms induce apoptosis of HL-60 leukemia cells. *Journal of Agricultural and Food Chemistry*, 54, 1222–1229.
- Pulido, R., Bravo, L., & Saura-Calixto, F. (2000). Antioxidant activity of dietary polyphenols as determined by a modified ferric reducing antioxidant power assay. *Journal of Agricultural and Food Chemistry*, 48, 3396–3402.
- Re, R., Pellegrini, N., Proteggente, A., Pannala, A., Yang, M., & Rice-Evans, C. (1999). Antioxidant activity applying an improved ABTS radical cation decolorization assay. *Free Radicals. Biology & Medicine*, 26, 1231–1237.
- Reed, J. D., McDowell, R. T. E., Van Soest, P. J., & Horvath, P. R. J. (1982). Condensed tannins: A factor limiting the use of cassava forage. *Journal of the Science of Food and Agriculture*, 33, 213–220.
- Ribeiro, J. C., Antunes, L. M. G., Aissa, A. F., Darin, J. D. C., Veridiana Rosso, V., Mercadante, A. Z., et al. (2010). Evaluation of the genotoxic and antigenotoxic effects after acute and subacute treatments with açai pulp (*Euterpe oleracea* Mart.) on mice using the erythrocytes micronucleus test and the comet assay. *Mutation Research*, 695, 22–28.
- Rufino, M. S. M., Alves, R. E., Brito, E. S., Pérez-Jiménez, J., Saura-Calixto, F., et al. (2010). Bioactive compounds and antioxidant capacities of 18 non-traditional tropical fruits from Brazil. *Food Chemistry*, 121, 996–1002.
- Rufino, M. S. M., Alves, R. E., Brito, E. S., Pérez-Jiménez, J., & Saura-Calixto, F. D. (2009). Total phenolic content and antioxidant activity in acerola, açai, mangaba and uvaia fruits by DPPH method. *Acta Horticulturae*, 841, 459–462.
- Rufino, M. S. M., Alves, R. E., Brito, E. S., Silveira, M. R. S., & Moura, C. F. H. (2009). Quality for fresh consumption and processing of some non-traditional tropical fruits from Brazil. *Fruits*, 64, 361–370.
- Rufino, M. S. M., Fernandes, F. A. N., Alves, R. E., & Brito, E. S. (2009). Free radical-scavenging behaviour of some north-east Brazilian fruits in a DPPH* system. *Food Chemistry*, 114, 693–695.
- Sánchez-Alonso, I., Jiménez-Escrig, A., Saura-Calixto, F., & Borderías, A. J. (2006). Effect of grape antioxidant dietary fiber on the prevention of lipid oxidation in minced fish: Evaluation by different methodologies. *Food Chemistry*, 101, 372–378.
- Sánchez-Moreno, C., Larrauri, J. A., & Saura-Calixto, F. (1998). A procedure to measure the antiradical efficiency of polyphenols. *Journal of the Science of Food and Agriculture*, 76, 270–276.
- Saura-Calixto, F. (1998). Antioxidant dietary fiber product: A new concept and a potential food ingredient. *Journal of Agricultural and Food Chemistry*, 48, 4303–4306.
- Saura-Calixto, F., García-Alonso, A., Goñi, I., & Bravo, L. (2000). In vitro determination of the indigestible fraction in foods: An alternative to dietary fiber analysis. *Journal of Agricultural and Food Chemistry*, 48, 3342–3347.
- Schauss, A. G., Wu, X., Prior, R. L., Ou, B., Patel, D., Huang, D., et al. (2006). Phytochemical and nutrient composition of the freeze-dried amazonian palm berry, *Euterpe oleracea* Mart. (Acai). *Journal of Agricultural and Food Chemistry*, 54, 8598–8603.
- Schreckinger, M. E., Lotton, J., Lila, M. A., & Mejia, E. G. (2010). Berries from South America: A comprehensive review on chemistry, health potential, and commercialization. *Journal of Medicinal Food*, 13, 233–246.
- Scott, R. W. (1979). Colorimetric determination of hexauronic acids in plant materials. *Analytical Chemistry*, 51, 936–941.
- Serrano, J., Goñi, I., & Saura-Calixto, F. (2007). Food antioxidant capacity determined by chemical methods may underestimate the physiological antioxidant capacity. *Food Research International*, 40, 15–21.
- Singleton, V. L., Orthofer, R., & Lamuela-Raventós, R. M. (1999). Analysis of total phenols and other oxidation substrates and antioxidants by means of Folin-Ciocalteu reagent. *Methods in Enzymology*, 299, 152–178.
- Southgate, D. A. (1969). Determination of carbohydrates in foods. Unavailable carbohydrates. *Journal of the Science of Food and Agriculture*, 20, 331–335.
- Souza, M. C., Figueiredo, R. W., Maia, G. A., Alves, R. E., Brito, E. S., Moura, C. F. H., et al. (2009). Bioactive compounds and antioxidant activity on fruits from different açai (*Euterpe oleracea* Mart) progenies. *Acta Horticulturae*, 841, 455–458.
- Spiller, G. A. (2001). *Handbook of dietary fiber in human nutrition*. Boca Raton: CRC Press.
- Wong, J. M., Souza, R., Kendall, C. W., Emam, A., & Jenkins, D. J. (2006). Colonic health: Fermentation and short-chain fatty acids. *Journal of Clinical Gastroenterology*, 40, 235–243.