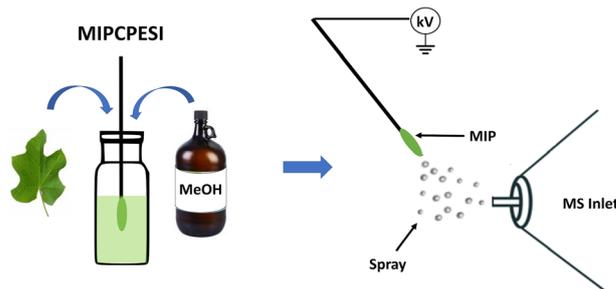


# Molecularly Imprinted Polymer-Coated Probe Electrospray Ionization Mass Spectrometry Determines Phorbol Esters and Deoxyphorbol Metabolites in *Jatropha curcas* Leaves

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**Abstract.** In this study, a molecularly imprinted polymer-coated probe electrospray ionization mass spectrometry (MIPCPEI-MS) method was developed for detection of phorbol esters (PEs) and deoxyphorbol metabolites in *Jatropha curcas* leaves. Such an approach was established by sticking on a metallic needle a molecularly imprinted polymer to particularly design a MIP-coated probe for selective sampling and ionization of PEs and

deoxyphorbol metabolites. By a subsequent application of a high voltage and methanol, as spray solvent, ESI was generated for direct and rapid analysis under ambient and open-air conditions. MIP-coated probe exhibited a high sampling capacity of the PEs and its metabolites in methanolic extracts of *J. curcas* leaves compared with the non-imprinted polymer (NIP)-coated probe. MIPCPEI-MS allowed the detection of phorbol 12,13-diacetate (PDA) from *J. curcas* leaves with minimal sample preparation, and with detection limit and quantification reaching 0.28  $\mu\text{g/mL}$  and 0.92  $\mu\text{g/mL}$ , respectively. Also, good linearity was obtained with  $R^2 > 0.99$  and precision and accuracy values between 4.06–13.49% and –1.60 to –15.26%, respectively. The current method was successfully applied to screening methanolic extracts of six different *J. curcas* leaf genotypes (three toxic and three non-toxic). PDA and three PE deoxyphorbol metabolites were identified only from toxic genotypes, in which PDA was determined with concentration ranging from  $222.19 \pm 23.55$  to  $528.23 \pm 19.72$   $\mu\text{g/g}$ . All these findings support that the MIPCPEI-MS method developed here has a high potential for the analysis of PEs in plant extracts enabling differentiation of toxic and non-toxic genotypes earlier in the leaves.

**Keywords:** Probe electrospray ionization (PESI), Molecularly imprinted polymer (MIP), Phorbol esters

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## Introduction

*Jatropha curcas* L. is an oilseed plant found in tropical and subtropical regions [1]. Its seeds consist of 30–40% of oil

which makes it useful for biodiesel production. After the oil extraction from the seeds, a pinion cake biomass with high protein content (53–63%) is produced. This raw material has a high potential for application in animal feed [2]. However, due to the existence of phorbol esters (PEs) on pinion cake, its use for animal feed is not allowed. PEs are toxic compounds that promote tumors, platelet aggregation, apoptosis, cell differentiation, and other adverse metabolic effects [3].

The structure of PEs depends on the tetracyclic diterpene carbon skeleton. Hydroxylation reactions at different positions

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in this basic structure lead to the formation of different structures, as reported for several unstable species of diterpene esters identified from the oil of *J. curcas* seeds [4].

Researches have been conducted to reduce or eliminate the PEs' content from *J. curcas*. In Brazil, the Brazilian Agricultural Research Company (Embrapa) is using genetic breeding approaches to produce new *J. curcas* genotypes with the absence of toxic compounds, excellent productivity, resistant to pests and diseases, and well-adapted to country's growing conditions [5, 6]. Traditionally, the differentiation of toxic and non-toxic genotypes is done by detection of PEs in the seeds; therefore, it is necessary that the cultivar reaches the age of seed production. Also, due to the complexity of the samples, extraction and separation techniques are required. Thus, new analytical methods to determine the content of PEs and its metabolites in the unusual matrices as leaves instead of *J. curcas* seed and oil, in order to estimate if the new genotypes are toxic or non-toxic, are highly desirable. Such determination would be advantageous, saving time, because it would not be necessary to wait for the plant growth. For instance, the determination would be made early, in the leaves.

Mass spectrometry (MS) has been widely used in the *J. curcas* analysis for the qualitative and quantitative determination of PEs, as well as in the differentiation of toxic and non-toxic genotypes [7–9]. MS is one of the most sensitive analytical techniques with a wide range of applicability for qualitative and quantitative analysis [10]. As the MS measures only molecules and atoms charged, the ionization process plays an important role in MS [11]. Traditionally, the MS analyses involve ionization techniques under vacuum conditions such as electron ionization (EI), chemical ionization (CI), and matrix-assisted laser desorption ionization (MALDI), as well as atmospheric pressure ionization (API) techniques, such as electrospray ionization (ESI) and atmospheric pressure chemical ionization (APCI) [12]. However, to perform analysis of complex matrixes, sometimes, it is unavoidable the use of laborious sample preparation steps [10].

During the last decade, ionization techniques under ambient conditions have been developed, allowing direct MS analysis with minimum or no sample preparation [10, 12, 13]. Since the development of desorption electrospray ionization (DESI) and direct analysis in real time (DART), several dozens of new ambient ionization methods have been described [12–14], as extractive electrospray ionization (EESI), rapid evaporative ionization (REIMS), leaf spray, paper spray (PS), and probe electrospray ionization (PESI). All these new ionization methods have been established as prominent methods for rapid and direct analysis under ambient conditions.

Leaf spray is an ambient ionization method applicable to direct detect chemical constituents of intact plant materials [15]. However, the detection of PEs direct from the leaves can be very difficult, mostly due to the lower content of PEs, found to be in concentrations about 1/10 than in the seeds [8].

Hence, in our previous studies, leaf spray was not capable of directly detecting PEs from *J. curcas* leaves [7].

Among ambient ionization methods, PESI has recently attracted attention for being a method derived from ESI with broad applicability and low cost that can be easily assembled in any MS laboratory. PESI was originally developed for liquid sample analysis by Hiraoka et al. [16]. Briefly, a non-capillary needle is used as a sampling probe and ESI emitter under ambient conditions. The needle touches the sample for analyte extraction, and subsequently, the needle is positioned in front of the mass spectrometer inlet orifice. A high voltage is applied to the needle, and an electrolytic spray is formed, followed by the ion analysis by MS. PESI has already been employed for direct MS analysis of foods [17], psychoactive drugs [18], plants [19], metabolites from mouse liver [20], and brain [21], including detection of metabolites at subcellular levels [22].

Despite all advantages, PESI presents some limitations, especially in the analysis of complex samples (e.g., blood, urine, and plants extract), including ionization suppression, low sensitivity at trace level, and narrow dynamic range [12, 23, 24]. In addition, PESI-MS analysis is generally performed using an inert surface needle, which has no enrichment capability. In this regard, it is rather challenging to detect PEs directly and its metabolites from *J. curcas* leaves using the PESI-MS. So, to perform this task, a needle with a high analyte enrichment capability should be developed for PESI-MS.

Integration of sampling, analyte enrichment, and ionization into a single device has emerged as a strategy for highly sensitive analysis in complex matrices using ambient ionization techniques. To date, solid-phase microextraction (SPME)-based methods [25] as surface-coated wooden-tip ESI [26] and coated blade spray ionization [27] have been described using such approach for direct and sensitive analysis in complex samples.

Furthermore, polymers may be used as the substrate for MS analyses [28]. Another approach consists of combine molecularly imprinted polymers (MIPs) as an adsorbent to sampling, extraction, and analyte enrichment with ambient ionization methods [14, 29, 30]. MIPs are materials that can provide a chemically selective surface for analyte enrichment. MIP synthesis generates imprinted sites, with similar size and shape, to a target analyte, which are prepared from cross-linking monomers and functional monomers in the presence of a target molecule, known as template [31, 32]. MIPs have been employed in combination with many ambient ionization methods. For example, MIP was coupled to natural ambient sonic-spray ionization (EASI-MS) for analysis of antipsychotic drugs from urine [33], to wooden-tip electrospray [34] for detection of trace malachite green in aquatic products as well to DESI-MS and paper spray ionization (PSI-MS) for analysis of pesticides from water [35] and fruits [14], respectively.

To improve PESI-MS performance in terms of sensitivity and selectivity, we describe the first example, to the best of our

knowledge, of using a molecularly imprinted polymer-coated probe electrospray ionization mass spectrometry method for selective analysis of PEs and its metabolites in *J. curcas* leaf extract. The developed probe, also, to being cost-effective and portable has imprinting cavities of MIPs with specific sites to binding PEs and its metabolites with selective enrichment even at trace levels. Determination of PEs and a series of three deoxy-phorbol metabolites using toxic and non-toxic genotypes of *J. curcas* were realized.

## Experimental

### *Materials and Reagents*

Leaves of different *J. curcas* genotypes were provided by Embrapa Agroenergy (Brasília, DF). The toxic genotypes were identified as 113F, 199F, and 259F, and the non-toxic genotypes as 169F, 170F, and 183F. High-purity chromatographic grade methanol and ethanol were purchased from J. T. Baker (Philipsburg, USA). Formic acid, phorbol-12-myristate-13-acetate (PMA), phorbol 12,13-diacetate (PDA), sodium dodecyl sulfate (SDS), tetraethoxysilane (TEOS), and 3-aminopropyltriethoxysilane (APTES) were supplied by Sigma-Aldrich (São Paulo, Brazil). Hydrochloric acid (HCl) was obtained from Scharlab S. L. (Sentmenat, Spain). Deionized water was purified on a Gehaka water-purification system (São Paulo, Brazil). Stained steel rods (4-cm length and diameter of 0.6 mm) and resin epoxy were purchased from the local market.

### *Preparation of Plant Extract*

The leaves were minced into smaller pieces. In sequence, 100 mg of the leaves was placed in an Eppendorf tube with 1 mL of methanol. The tube was placed into an ultrasonic bath. Extraction was carried out in ambient temperature for 10 min. This extraction procedure was repeated two times for each sample to reach 3 mL of methanolic extract, which was filtered through filter paper before MS analysis.

### *Preparation and Characterization of MIP- and NIP-Coated Probe*

For MIP synthesis, 5.9 mmol of TEOS, 12.8 mmol of water, 0.03 mmol of HCl, and 6.8 mmol of ethanol were added to a Falcon tube. The mixture was stirred for 3 min in a vortex and allowed to stand for 24 h. In sequence, 1.4 mmol of APTES, 3.5 mmol of SDS, and 616.2 mmol of the phorbol 12-myristate 13-acetate were added in another Falcon tube. The mixture was stirred for 3 min in a vortex, added to the mixture prepared above, and allowed to stand for 24 h.

Posteriorly, the PMA used as template was removed with methanol by Soxhlet extraction using a Sebelin heating battery from Ethiktechnology (São Paulo, Brazil) at 70 °C for 48 h. To perform the solvent extraction, the imprinted polymers were placed in a paper thimble inside the Soxhlet apparatus; clamps were used to fix the Soxhlet apparatus to the condenser and the extraction flask containing 200 mL of methanol. After solvent

extraction, the polymers were put in the oven at 50 °C to dry for 6 h; then, they were powdered using porcelain mortar and pestle.

The non-molecularly imprinted polymer (NIP) was also synthesized for purposes of comparing the morphology and performance of the polymers as the extracting phase of the compounds of interest. The NIP synthesis occurred similarly to MIP synthesis without the addition of the template molecule.

Finally, the powdered MIP and NIP were stuck on stained steel needles with 4 cm length and 0.6 mm of diameter using a resin epoxy Loctite from Henkel (São Paulo, Brazil). To ensure the evenness, 1 cm length of the needles were covered with 15 mg of the polymers. The morphology of the MIP and NIP were characterized via scanning electron microscopy (SEM) analysis using a JEOL JSM-6610 (Tokyo, Japan).

### *Sampling, Extraction, and Mass Spectrometry Analysis*

Mass spectra were acquired on a Q Exactive™ Hybrid Quadrupole-Orbitrap mass spectrometer (Thermo Scientific, San Jose, USA) and LCQ Fleet (3D ion trap) mass spectrometer (Thermo Scientific, San Jose, USA), recording an appropriate  $m/z$  in the positive ion mode. Accurate mass measurement was performed using positive ion electrospray Q Exactive Orbitrap with 70,000 resolution power at  $m/z$  200 using the following parameters: spray voltage of 3.5 kV, the capillary temperature of 275 °C, and S-lens RF level of 50%.

For extractive sampling using the MIPCPEI, about 1 cm of the coated probe was immersed in the methanolic extracts under magnetic stirring for 10 min. After sampling extraction, MIPCPEI probe was washed with ultra-purified water to eliminate other matrix components and then drying at 40 °C for 30 min.

MIP and NIPCPEI-MS analyses were performed using the LCQ Fleet mass spectrometer (Thermo Scientific, San Jose, USA). MIPCPEI probe was positioned 5 mm away from the MS inlet at an angle of 105°. To perform ESI, methanol with 0.1% formic acid was delivered at a flow rate 15  $\mu\text{L}/\text{min}$  into MIP-coated surface of the probe, which was connected with a high voltage of 3.5 kV available from the MS. Mass spectra were acquired in positive ion mode with a capillary temperature of 250 °C. Structure confirmation of analytes was performed by tandem mass spectrometry (MS/MS) running collision-induced dissociation (CID) of 20 eV. Helium gas (99.99 purity) was used as the collision gas in the CID experiments.

### *Analytical Performance*

To PEs quantification, non-toxic *J. curcas* leaf extracts were chosen as matrix, and it was spiked with PDA at different concentration levels (1, 5, 10, 15, 20, 25, and 30  $\mu\text{g}/\text{mL}$ ). Selected reaction monitoring (SRM) mode was used for the quantification experiments (transition of  $m/z$  431 to 391). The calibration curve was plotted using the SRM peak about the analyte concentration. The theoretical limit of detection (LOD), representing the smaller amount of the analyte that can be detected, and the limit of quantification (LOQ), representing

the smaller amount of the analyte that can be determined with acceptable precision and accuracy, were determined from the calibration curve. Theoretical LOD was determined by the following equation  $S_{\text{blank}} + 3 \times \sigma_{\text{blank}}$ , and LOQ was obtained by  $S_{\text{blank}} + 10 \times \sigma_{\text{blank}}$  where  $S_{\text{blank}}$  is the average blank signal and  $\sigma_{\text{blank}}$  is the standard deviation of the signal from 10 replicates. Analyses to determine precision, accuracy, and recovery were performed intraday and interday with measurements made in triplicate using three concentrations (5, 15, and 25  $\mu\text{g/mL}$ ) within the linearity range. Precision was calculated by the equation  $\text{CV} = (\text{SD}/\text{AC}) \times 100$ , where CV is the coefficient of variation, SD is the standard variation, and AC is the analyzed concentration. Accuracy was calculated by the formula  $E = [(\text{AC} - \text{NC})/\text{NC}] \times 100$ , where  $E$  is the relative error and NC is the nominal concentration [14, 36]. The percentage recovery,  $R$ , was calculated by the formula  $R = (\text{AC}/\text{NC}) \times 100$ . Finally, the developed MIPCPEI-MS method was applied for quantification of the phorbol 12,13-diacetate in samples from six different *J. curcas* genotypes.

## Results and Discussion

### *J. curcas* Leaf Extract Characterization

Routinely, PEs are determined on seeds of *J. curcas*. However, to enable protein-rich *J. curcas* pinon cake in animal feed is necessary to evaluate PEs and its metabolites in other matrices, as the leaves of *J. curcas*. Such determination would be useful, mainly to save time, on researches that attempt to produce non-toxic genotypes of *J. curcas*. In this way, methanolic extracts from *J. curcas* leaves were analyzed via ESI + Q Exactive mass spectrometer to identify the profile of phorbol esters and its metabolites. As displayed in Fig. 1S, only one PE were identified. The phorbol 12,13-diacetate (PDA) was detected as protonated ion  $[\text{M} + \text{H}]^+$  of  $m/z$  449.21460 in all three toxic genotype leaf extracts. Besides the detection of PDA, as displayed in Fig. 2S, a set of deoxyphorbol compounds, derived from phorbol [37], were detected as protonated ions  $[\text{M} + \text{H}]^+$ . These compounds include the 12-deoxyphorbol (I) of  $m/z$  347.18530, 4,12-dideoxyphorbol (II) of  $m/z$  329.17474, and 4,9,12-trideoxyphorbol (III) of  $m/z$  311.16417.

Phorbol esters are unstable molecules that commonly lose  $\text{CH}_3\text{COOH}$  and  $\text{H}_2\text{O}$  groups as reported by Vogt et al. [9]. Fig. 3S illustrates a mechanistic proposal explaining the formation of deoxyphorbol derivatives from the phorbol, by removal of oxygen atoms from the parent polyol. However, it remains undetermined if these phorbol analogs are derived from dehydration route or transannular cyclization of less oxygenated lathyrene precursors [38].

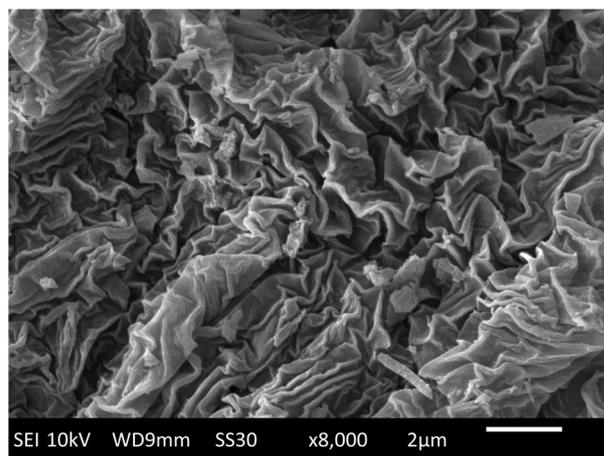
### MIPCPEI Probe Characterization

The MIP synthesis includes the formation of the complex between the monomer and the template molecule followed by template removal creating imprinting sites on the polymer surface [32]. Here, aminopropyl triethoxysilane was used as

the monomer, tetraethoxysilane as crosslinker reagent, and phorbol 12-myristate 13-acetate was employed as a template in MIP synthesis for PE analysis. The choice of PMA as template was a strategic one that no peaks related to this molecule were detected in the samples previously analyzed by ultra-high-resolution mass spectrometry.

Morphological characterization of the synthesized polymers was performed via scanning electron microscopy (SEM) analysis as illustrated in Figure 1. Note the presence of macroporous layer structure, which covers the stained steel needle surface. According to Cormack and Elorza [39], macroporous polymers are mechanically stable and generally preferable for MIP synthesis. As shown in Figure 1(a), the molecularly imprinted polymer has a higher roughness surface with high porosity and imprinting cavities compared with the non-imprinted polymer stated in Figure 1(b). These results are in agreement with the characterization from other MIPs

(a)



(b)

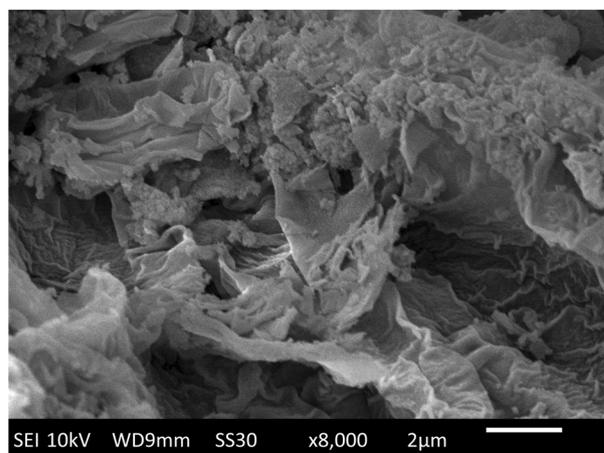


Figure 1. SEM image ( $\times 8000$ ) of molecularly imprinted polymer (a) and non-imprinted polymer (b)

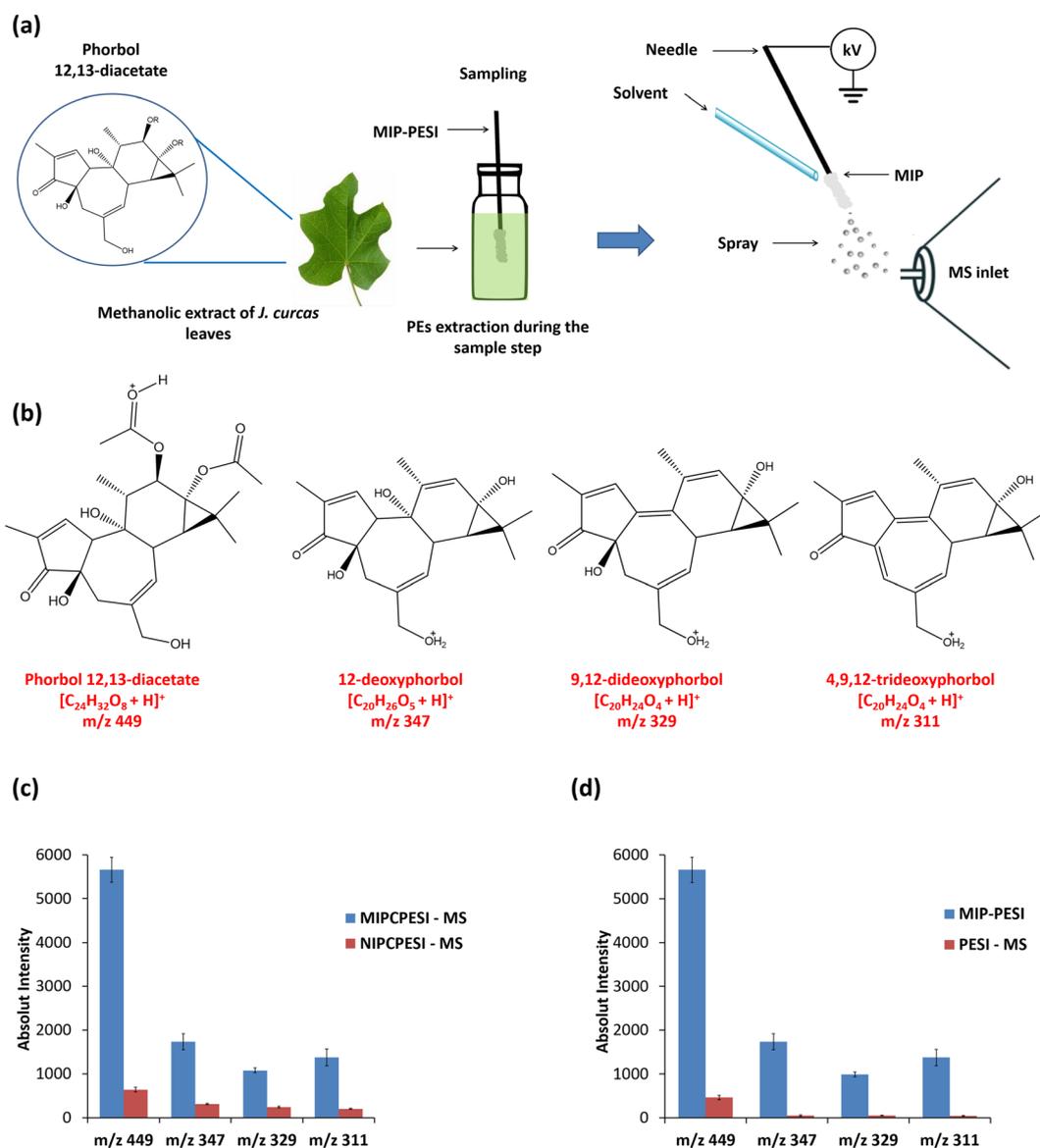
synthesized as extraction phase [14, 40, 41].

### MIPCPESI-MS Analysis

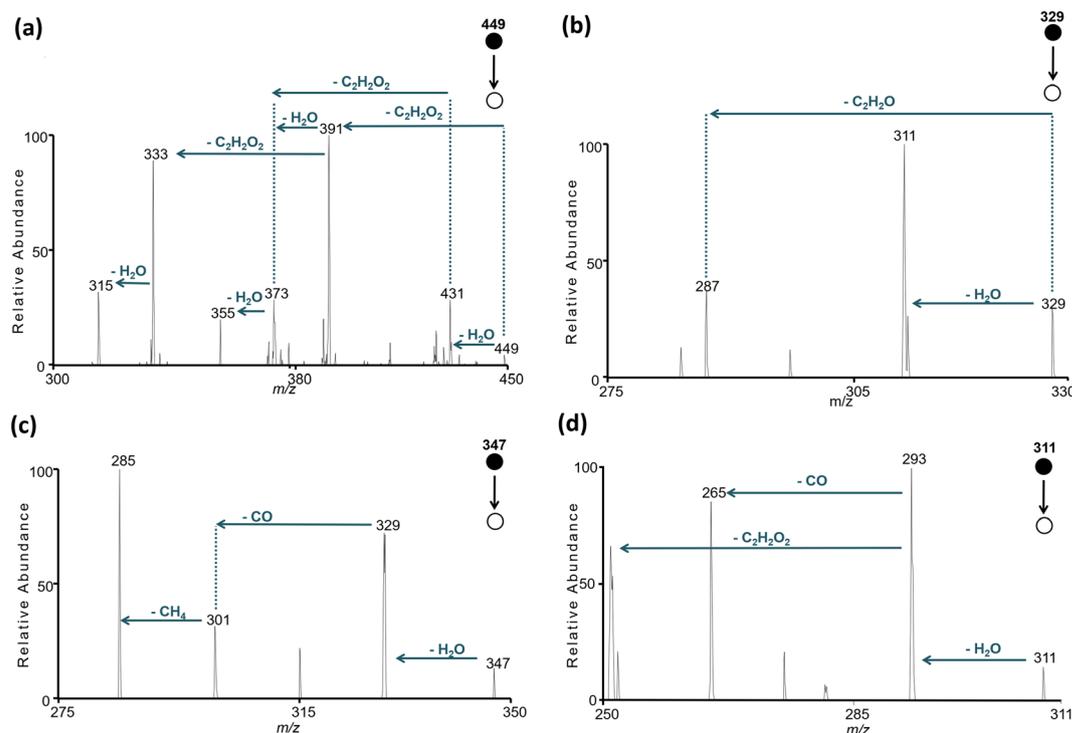
As PEs and its phorbol derivatives (e.g., deoxyphorbol) have a common diterpene nucleus which contained different content of hydroxy groups esterified or not, the selective extraction of MIPS on stained steel needle is based on specific sites corresponding to both diterpenic ester and hydroxy group. Methanolic extracts were prepared using 100 mg of toxic *J. curcas* genotype leaves to investigate the analytical performance of MIPCPESI probe in terms of analyte enrichment. The extracted MIPCPESI probe was placed in front of the MS inlet, and 3.5 kV was applied to the probe.

Methanol with 0.1 formic acids was delivered into MIP-coated surface of the probe at a flow rate 15  $\mu\text{L}/\text{min}$ , as displayed in Figure 2(a). Interestingly, the PDA and the three deoxyphorbol compounds described earlier were identified using MIPCPESI-MS. Figure 2(b) displays the structural formula of detected compounds. As shown in Figure 2(c), the extractive capacity of the MIPCPESI probe was evaluated employing the comparison of the signal intensities obtained in the mass spectra for PDA and three deoxyphorbol detected.

These results demonstrate that the adsorption of the diterpenic phorbol ester group (PDA) is much higher as compared to deoxyphorbol, which contain only diterpenic hydroxy groups (compounds I, II, and III). The data hence suggest that the MIP of MIPCPESI probe is primarily extracting



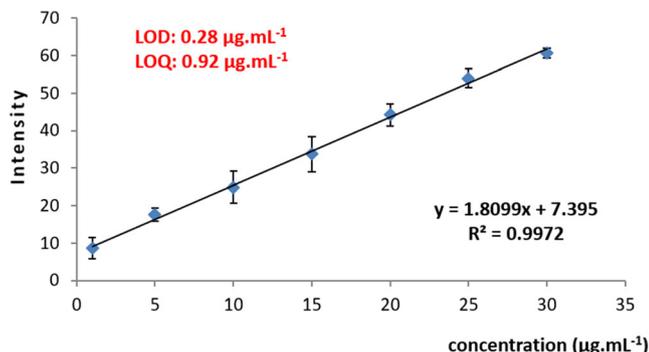
**Figure 2.** Schematic of the experimental procedure of MIPCPESI-MS analysis (a); structures of the detected phorbol 12,13-diacetate, 12-deoxyphorbol, 9,12-dideoxyphorbol, and 4,9,12-trideoxyphorbol (b); bar graph comparing the signal intensities obtained from mass spectra of detect PEs using MIPCPESI-MS and NIPCPESI-MS (c); and comparison between ionization performance of MIPCPESI-MS and traditional PESI-MS (d). Errors bars show the standard deviation of triplicate measurements



**Figure 3.** MIPCPESI + MS/MS spectra of the  $[M + H]^+$  ions for phorbol 12,13-diacetate (a), 12-deoxyphorbol (b), 4,12-deoxyphorbol (c), and 4,9,12-trideoxyphorbol (d)

compounds with diterpenic phorbol moiety, mainly PDA. This experimental result was expected due to the more significant similarity of the PDA to the PMA, used as a template for design imprinting sites in polymer, compared with the similarity between PMA and deoxyphorbol compounds. MIP selectivity is based on its ability to recognize and re-bind the template molecule [31, 42, 43].

The comparison between the NIP and MIPCPESI-MS analysis in Figure 2(c) corroborates with a previously discussed statement. MIPCPESI-MS provides higher signal intensities



**Figure 4.** The calibration curve, theoretical LOD, and LOQ obtained via MIPCPESI + MS/MS analyses of the blank matrix with PDA addition in concentrations ranging 1 to 30  $\mu\text{g mL}^{-1}$ . Error bars show the standard deviation of measurements made in triplicate

for the peaks related to phorbol ester and deoxyphorbol compounds in comparison with NIPCPESI-MS. Imprinted cavities of MIPCPESI probe favors PE and phorbol analog extraction, leading us to improve the selective and specific extraction of such compounds at trace levels from *J. curcas* leaf extract.

Fig. 4S displays the positive ion mode mass spectra of methanolic extracts of toxic *J. curcas* genotype leaves obtained by MIPCPESI probe and by the conventional PESI probe (stained steel needle). As discussed above, MIPCPESI probe allowed the detection of PDA and phorbol analogs and deoxyphorbol compounds. On the other hand, when the methanolic extract of toxic *J. curcas* leaves was interrogated by conventional PESI probe, only PDA ( $m/z$  449) was detected at low abundance. The ions of  $m/z$  347, 329 and 311, referent to deoxyphorbol compounds I, II, and III, respectively, were not detected. As mentioned earlier, the plant extract is a complex matrix, and its composition might cause a deleterious effect on the ionization process. These components can suppress analyte signal, causing it to be cloaked by interferent ions, thus equally lowering sensitivity and complicating the spectrum. As MIPCPESI probe integrates both analyte enrichment and ionization capability, it is less susceptible to ionic suppression inherent of the matrix sample compared with the non-coated stained steel needle used in conventional PESI. As noted in the bar chart shown in Figure 2(d), MIPCPESI probe possessed excellent extraction capacity for analysis of methanolic *J. curcas* leaf extract compared to conventional PESI.

**Table 1.** Precision, Accuracy, and Recovery for MIPCPEI Method in the Analysis of Phorbol 12,13-Diacetate in *J. curcas* Leaf Extracts

|          |   |                 |                  |                  |
|----------|---|-----------------|------------------|------------------|
| Intraday | NC <sup>a</sup> ( $\mu\text{g mL}^{-1}$ )                           | 5               | 15               | 25               |
|          | AC <sup>b</sup> ( $\mu\text{g mL}^{-1}$ ) $\pm$ UC <sup>c</sup> (%) | 4.92 $\pm$ 1.12 | 13.24 $\pm$ 1.72 | 27.68 $\pm$ 4.37 |
|          | Precision (CV <sup>d</sup> %)                                       | 13.49           | 7.72             | 9.37             |
|          | Accuracy E <sup>e</sup> %   | -1.60           | -11.70           | 10.73            |
|          | Recovery %  | 98.40           | 88.27            | 110.72           |
| Interday | NC <sup>a</sup> ( $\mu\text{g mL}^{-1}$ )                           | 5               | 15               | 25               |
|          | AC <sup>b</sup> ( $\mu\text{g mL}^{-1}$ ) $\pm$ UC <sup>c</sup> (%) | 5.65 $\pm$ 0.74 | 12.71 $\pm$ 0.93 | 26.41 $\pm$ 1.81 |
|          | Precision (CV <sup>d</sup> %)                                       | 7.78            | 4.35             | 4.06             |
|          | Accuracy E <sup>e</sup> %   | 12.77           | -15.26           | 5.65             |
|          | Recovery %  | 113.00          | 84.73            | 105.64           |

<sup>a</sup>Nominal concentration<sup>b</sup>Analyzed concentration<sup>c</sup>Expanded uncertainty ( $k = 2$  for  $P = 95\%$ )<sup>d</sup>Coefficient of variation<sup>e</sup>Relative error

MIPCPEI-MS is suitable for the determination of trace level of PEs and phorbol analogs in *J. curcas* leaves.

As MIPCPEI mass spectra were acquired on LCQ mass spectrometer, PDA and deoxyphorbol ions were subjected to the collision-induced dissociation (CID) experiments to confirm their identities. Figure 3 shows the MIPCPEI + MS/MS spectra of  $[M + H]^+$  ions for PDA and deoxyphorbol molecules.

In Figure 3(a), CID of the  $m/z$  449 corroborates to phorbol 12,13-diacetate, exhibiting characteristics loss of water (18 Da) molecules and acetate groups present in their structure observed in different fragmentation routes. The PDA molecule has several ionization sites; therefore, many fragmentation routes are possible. Fig. 5S shows some possible fragmentation routes from ionization in one of the esters groups (a) and ionization in one of the hydroxyl groups (b).

In CID spectra, the 12-deoxyphorbol, 4,12-deoxyphorbol, and 4,9,12-deoxyphorbol, respectively shown in Figure 3(a), (b), and (c), exhibited neutral losses of H<sub>2</sub>O (18 Da) molecules. The dihydroxylation is commonly observed in phorbol structures [4], as observed in CID spectra; the elimination of water molecules brings stability and abundance to the resulting fragment ions. In addition to removal of hydroxyl groups, for the  $m/z$  347 occurs the elimination of the ketone group (C=O, 28 Da) (see fragmentation mechanism proposed in Fig. 6S). For the 4,12-dideoxyphorbol of  $m/z$  329, the elimination of one water molecule generates the ion of  $m/z$  311 commonly found in samples containing PEs since it is related to the main degradation product of the phorbol nucleus [8, 9]. In a second fragmentation route, via a rearrangement [1, 3], H occurs the elimination of

ethanol group (HC  $\equiv$  COH, 42 Da). The proposed fragmentation mechanisms from ionization in two different sites are shown in Fig. 7S. In CID spectrum, the ion 4,9,12-trideoxyphorbol ( $m/z$  311) exhibit typical fragmentation pattern obtained by removal of a water molecule followed by the loss of a ketone group consisting with the commonly reported in the literature (see fragmentation mechanism proposed in Fig. 8S).

### MIPCPEI Probe Analytical Performance

Quantitative analysis of the phorbol 12,13-diacetate ( $m/z$  449) in methanolic extracts of *J. curcas* leaves achieved a calibration curve constructed with samples of the blank matrix with addition of PDA standard in seven concentrations ranging from 1 to 30  $\mu\text{g/mL}$ . According to Baldini et al. [8], the expected concentration of PEs in *J. curcas* leaves is approximately 525  $\mu\text{g/g}$ . Therefore, the working range was chosen for the 100 mg of leaves used in the preparation of 3 mL of samples in which an aliquot of 1 mL was analyzed [8]. The SRM (single reaction monitoring) mode was used to monitor the transition of  $m/z$  449 to 391 in the function of the concentration of PDA standard in the samples. Fragmentation spectra used to build the analytical curve are shown in Fig. 9S.

Using MIPCPEI-MS, the signal intensities ratios of the most abundant MS/MS transitions were found to be linear ( $y = 1.8099x + 7.395$ ,  $R^2 = 0.9972$ ). Theoretical LOD and LOQ were calculated from the calibration curve, resulting, respectively, in values of 0.28 and 0.92 ppm as shown in Figure 4. The LOD and LOQ values found in the  $\mu\text{g/mL}$  range are in

**Table 2.** Concentrations of Phorbol 12,13-Diacetate Determined in *J. curcas* via Analysis MIPCPEI + MS/MS

| Sample | PDA concentration ( $\mu\text{g mL}^{-1}$ ) $\pm$ UC <sup>a</sup> (%) |
|--------|---|
| 113F   | 344.85 $\pm$ 20.27  |
| 199F   | 222.19 $\pm$ 23.55  |
| 259F   | 528.23 $\pm$ 19.72  |
| 170F   | < LOD <sup>b</sup>  |
| 171F   | < LOD <sup>b</sup>  |
| 183F   | < LOD <sup>b</sup>  |

<sup>a</sup>Expanded uncertainty ( $K = 2$  for  $P = 95\%$ )<sup>b</sup>LOD = 0.28  $\mu\text{g g}^{-1}$

agreement with results reported in the literature for PEs' determination in leaves and seeds of *J. curcas* via analyses HPLC-MS/MS and HPLC-UV [8, 44].

Analyses to determine precision and accuracy were done in triplicate at low (5 ppm), medium (15 ppm), and high (25 ppm) concentration levels contemplating the linearity range. For intraday analyses, recovery values between 88.27 and 110.72% are found, also, the lowest precision value was 7.72% for the sample at 15 µg/mL, and the lowest accuracy value was -1.60% at 5 µg/mL. For interday analyses, the lowest values were obtained for the sample at 25 µg/mL of 4.35% for precision and 5.65% for accuracy, while the recovery was found between 84.73 and 113.00%. Table 1 shows all precision and accuracy values obtained for MIPCPEI method. The presented coefficients of variation and relative errors lower than 20% are established in the Brazilian Manual of the Ministry of Agriculture, Livestock, and Supply (MAPA). Moreover, these results with precision and accuracy close to 15% are in agreement with those reported in the literature using MIP and ambient mass spectrometry [14, 33].

Posteriorly, MIPCPEI method was employed in the determination of phorbol 12,13-diacetate in methanolic extracts of *J. curcas* leaves. The genotypes analyzed were yielded by Embrapa Agroenergia (Brasília, Brazil). The results reported in Table 2 show the content of PDA determined in toxic (113F, 199F, and 259F) and non-toxic (170F, 171F, and 183F) genotypes. The sample 259F presented the highest concentration of PDA, and the lowest concentration was found for samples 199F. No peak related to phorbol esters were detected in non-toxic leaves.

## Conclusions

In this work, the practicality of the ionization by probe electrospray was combined with the selectivity of the MIP used as extraction phase in the determination of PEs in leaf extracts of *J. curcas*. The MIP-coated probe proved to be adequate for extraction of the PEs in methanolic extracts of *J. curcas* leaves with better performance as extraction phase in comparison with the coated non-imprinted polymer. We have demonstrated MIPCPEI as an effective method to perform an extraction with lower solvent demand and effective ionization for detection of a target molecule in a complex matrix using an ambient mass spectrometry technique. The method presented linearity with  $R^2$  greater than 0.99, theoretical LOD and LOQ in the ppm range, and satisfactory values of accuracy and precision lower than 20%. The method was applied to determine the contents of PDA in methanolic extracts of *J. curcas* leaves; no content of PEs was detected in non-toxic genotypes; for the toxic, concentrations ranging from  $222.19 \pm 23.55$  to  $528.23 \pm 19.72$  µg/g were determined. Therefore, the presented method is promising to be optimized and applied for other analytes and matrices.

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